

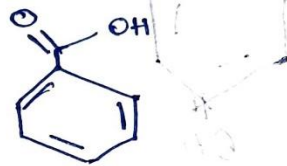
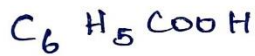
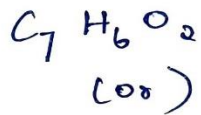


CLASS II PRESEERVATIVES

class (ii) Preservatives

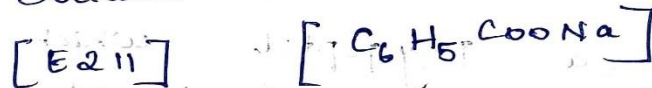
(i) Benzoic acid & salts

• Benzoic acid - antimicrobial agent



Aromatic Carboxylic acid

↳ Sodium Benzoate - soda



↳ Toothpaste, jam, beverage,

jelly, margarine, carbonated

beverages, fruit salad, pickles,

fruits, juices as antimicrobial

agent. [0.05% to 0.1%] concentration limit

[effective in acidity] is less pH i.e. increase

↳ pH 2.5 - 4.0

ESTERS: [of Benzoic acid]

↳ of p-hydroxybenzoic acid, methyl paraben, propyl paraben

↳ effective at higher pH  
↳ Antifungal & active against yeasts.

Esters of p-ott benzoic acids [synthetic parabens]

↳  $C_3H_7$  - (Propyl paraben)

• White crystals

• Melting point  $96.97^\circ C$

• Soluble in 2000 parts of water  
i.e., 1g in 2 litres of water

• Freely soluble in alcohol, either slightly in boiling water.

↳  $CH_3$  - (Methyl paraben)

• White needles

• Melting point  $131^\circ C$

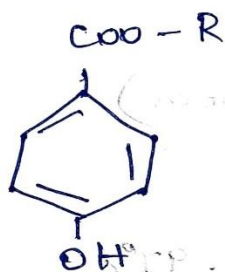
• Boiling point  $275 \pm 5^\circ C$  with decomposition

• 1g dissolves in 100ml of water, 10ml of warm water

To ml of warm glycerine

• Freely soluble in alcohol, acetone, ether

• Solubility in water  $\Rightarrow$  0.25% (at 20°C) (w/w)  
 $\Rightarrow$  0.30% (w/w) at 25°C



Common structure of parabens

where, R can be either  $\text{CH}_3$  or  $\text{C}_2\text{H}_5$  etc.

## Anti-oxidants

### Antibiotics

• very few are permitted

• Nisin [poly peptide]

• produced by certain strains of *Streptococcus lactis*

• occurs naturally in milk & cheddar cheese

• Non-toxic, heat stable