

# WAVE FUNCTION

A variable quantity which characterizes de Broglie waves is known as **wave function** and is denoted by the symbol  $\Psi$ . The value of the wave function associated with a moving particle at point  $(x,y,z)$  and time 't' gives the probability of finding the particle at that time and at that point.

## PHYSICAL SIGNIFICANCE OF WAVE FUNCTION

- The wave function has **no physical meaning**.
- It is a **complex quantity** representing the matter **wave of a particle**.
- $|\psi|^2$  is **real and positive**, amplitude may be positive or negative but the intensity (square of amplitude) is always real and positive.
- $|\psi|^2$  represents the probability density or probability of **finding the particle in the given region**.
- For a given volume  $d\tau$ , probability  $P = \iiint |\psi|^2 d\tau$  where  $d\tau = dx dy dz$
- The **probability value** lies between **0 and 1**.
- $\iiint |\psi|^2 d\tau = 1$ , this wave function is called **normalized wave function**.

# SCHROEDINGER'S WAVE EQUATION

- Austrian scientist, **Erwin Schroedinger**
- describes the wave nature of a particle , derived in **mathematical form**
- connected the expression of **De-Broglie wavelength** with **classical wave equation**
- two forms of **Schroedinger's wave equation**
- Time **Independent** wave equation
- Time **dependent** wave equation

# SCHROEDINGER'S TIME INDEPENDENT WAVE EQUATION

□ Let us consider a system of stationary wave associated with a **moving particle**. Let  $\psi$  be the wave function of the particle along x, y and z coordinates axes at any time t.

The differential wave equation of a progressive wave with wave velocity 'u' can be written in terms of Cartesian coordinates as,

$$\frac{d^2\psi}{dx^2} + \frac{d^2\psi}{dy^2} + \frac{d^2\psi}{dz^2} = \frac{1}{v^2} \frac{d^2\psi}{dt^2} \quad \frac{\partial^2\psi}{\partial x^2} + \frac{\partial^2\psi}{\partial y^2} + \frac{\partial^2\psi}{\partial z^2} = \frac{1}{v^2} \frac{\partial^2\psi}{\partial t^2} \dots(1)$$

The solution of eqn(1) is  $\psi = \psi_0 e^{-i\omega t}$

Differentiating the above equation with respect to time 't' twice,

$$\frac{\partial\psi}{\partial t} = -i\omega\psi_0 e^{-i\omega t} = -i\omega\psi \quad \frac{\partial^2\psi}{\partial t^2} = -\omega^2\psi \dots(2)$$

Substituting (2) in (1)

$$\frac{\partial^2\psi}{\partial x^2} + \frac{\partial^2\psi}{\partial y^2} + \frac{\partial^2\psi}{\partial z^2} = \frac{-\omega^2\psi}{V^2}$$

# SCHROEDINGER'S TIME INDEPENDENT WAVE EQUATION

$$\omega = 2\pi\nu \quad \frac{\partial^2\psi}{\partial x^2} + \frac{\partial^2\psi}{\partial y^2} + \frac{\partial^2\psi}{\partial z^2} = \frac{-4\pi^2\nu^2\psi}{V^2}$$

substituting  $V = v\lambda$

$$\frac{\partial^2\psi}{\partial x^2} + \frac{\partial^2\psi}{\partial y^2} + \frac{\partial^2\psi}{\partial z^2} = \frac{-4\pi^2\nu^2\psi}{v^2\lambda^2} \quad \lambda = \frac{h}{mV}$$

$$\frac{\partial^2\psi}{\partial x^2} + \frac{\partial^2\psi}{\partial y^2} + \frac{\partial^2\psi}{\partial z^2} = \frac{-4\pi^2m^2V^2\psi}{h^2}$$

$$\text{total energy } E = \frac{1}{2}mV^2 + V$$

$$2m(E - V) = m^2V^2$$

$$\frac{\partial^2\psi}{\partial x^2} + \frac{\partial^2\psi}{\partial y^2} + \frac{\partial^2\psi}{\partial z^2} = \frac{-8\pi^2m(E - V)\psi}{h^2} \quad \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} = \nabla^2, \text{ where } \nabla \text{ is Laplacian Operator}$$

$$\nabla^2\psi = \frac{-8\pi^2m(E - V)\psi}{h^2} \Rightarrow \frac{-2m}{\hbar^2}(E - V)\psi \quad \text{since } \hbar = \frac{h}{2\pi}$$

$$\nabla^2\psi + \frac{2m}{\hbar^2}(E - V)\psi = 0 \text{ (Time independent Schrodinger wave eqn)}$$

# SCHROEDINGER'S TIME DEPENDENT WAVE EQUATION

The solution of the classical differential eqn. of wave system is,

$$\psi = \psi_0 e^{-i\omega t}$$

Differentiating the above equation with respect to time 't' ,

$$\frac{\partial \psi}{\partial t} = -i\omega \psi_0 e^{-i\omega t} = -i\omega \psi$$

Substituting  $E = h\nu$        $\omega = 2\pi\nu$

$$\frac{\partial \psi}{\partial t} = -i2\pi\nu \psi = -i2\pi \left( \frac{E}{h} \right) \psi$$

$$\frac{\partial \psi}{\partial t} = -i2\pi\nu \psi = -i \left( \frac{E}{\hbar} \right) \psi = -i^2 \left( \frac{E}{i\hbar} \right) \psi \quad \text{since } \hbar = \frac{h}{2\pi}$$

$$E\psi = i\hbar \frac{\partial \psi}{\partial t}$$

$$\nabla^2 \psi + \frac{2m}{\hbar^2} (E - V)\psi = 0 \text{ (Time independent Schrodinger wave eqn)}$$

$$\nabla^2 \psi + \frac{2m}{\hbar^2} E\psi - \frac{2m}{\hbar^2} V\psi = 0$$

# SCHROEDINGER'S TIME DEPENDENT WAVE EQUATION

$$\nabla^2 \psi + \frac{2m}{\hbar^2} i\hbar \frac{\partial \psi}{\partial t} - \frac{2m}{\hbar^2} V \psi = 0$$

multiply throughout by  $\frac{\hbar^2}{2m}$

$$\frac{\hbar^2}{2m} \nabla^2 \psi + E \psi - V \psi = 0$$

$$-\frac{\hbar^2}{2m} \nabla^2 \psi + V \psi = E \psi \text{ (Time dependent Schroedinger wave eqn)}$$

*The Hamiltonian operator*

$$H = -\frac{\hbar^2}{2m} \nabla^2 + V$$

$$H \psi = E \psi$$

*E = energy operator*